

UNIVERSITÀ DEGLI STUDI DI PALERMO

DEPARTMENT	Biomedicina, Neuroscienze e Diagnostica avanzata
ACADEMIC YEAR	2017/2018
BACHELOR'S DEGREE (BSC)	BIOMEDICAL LABORATORY TECHNIQUES
INTEGRATED COURSE	BIOCHEMISTRY AND PHYSICS - INTEGRATED COURSE
CODE	17558
MODULES	Yes
NUMBER OF MODULES	2
SCIENTIFIC SECTOR(S)	FIS/07, BIO/10
HEAD PROFESSOR(S)	DI GAUDIO FRANCESCA Ricercatore Univ. di PALERMO
OTHER PROFESSOR(S)	SAVOCA GAETANO Professore a contratto Univ. di PALERMO
	DI GAUDIO FRANCESCA Ricercatore Univ. di PALERMO
CREDITS	6
PROPAEDEUTICAL SUBJECTS	
MUTUALIZATION	
YEAR	1
TERM (SEMESTER)	1° semester
ATTENDANCE	Mandatory
EVALUATION	Out of 30
TEACHER OFFICE HOURS	DI GAUDIO FRANCESCA
	Friday 13:00 14:00 DIPARTIMENTO PROMOZIONE DELLA SALUTE, MATERNO INFANTILE DI MEDICINA INTERNA E SPECIALISTICA DI ECCELLENZA - PROMISE - "GIUSEPPE D'ALESSANDRO" - CQRC Via del Vespro n. 133, primo piano

PREREQUISITES	In order to be admitted to this Degree Course, students must pass a compulsory examination based on a general culture test that includes basic but basic topics. The course does not provide a course for chemistry either for the general part or for the organic chemistry absolutely propedeutic for Biochemistry. Therefore, to understand the themes of the Biochemistry Course, it is necessary for students to undertake a thorough in-depth study of at least organic chemistry.
LEARNING OUTCOMES	The student at the end of the course will acquire a good knowledge of the chemical and physical principles at the basis of human metabolism and laboratory tests that allow to evaluate metabolic function. Knowledge and understanding: The student must know and understand the molecular and chemical physical basis of matter, the characteristics of the elements and molecules and the principles underlying their chemical-physical behavior, the structure and properties of organic compounds And macromolecules of biological interest. Students will also need to know the biochemical and molecular basis of cellular activity and understand the mechanisms that regulate metabolic processes. In particular, they will need to know the molecules of biological interest and the main metabolic processes involved, know and understand the mode of action of the enzymes, the bases of their regulation and catalysis, the basics of bioenergetics and the general mechanisms of transduction Of extracellular signals.
ASSESSMENT METHODS	Learning of the student will be assessed through an oral examination at the end of the course. It will assess the acquired knowledge, processing and presenting capabilities. The evaluation is expressed in thirtieths. The threshold of sufficiency will be achieved when the student will have shown knowledge and understanding of the course topics, at least in general terms, and that they have a sufficient capacity to convey his knowledge to the examiner. The sufficiency threshold will be reached if the student shows knowledge and understanding of the issues at least in broad outline, and has application skills sufficient; he must also have presentation and argumentative skills allowing the transmission of his knowledge to the examiner. Below this threshold, the examination will be insufficient. The more the candidate will be able to interact with the examiner with his argumentative and presentation skills, and the more his knowledge and application capabilities will go into detail on the subjects under evaluation, the more the judgement will be positive. The evaluation is expressed using a 30-point scale. ECTS grades: A – A+ Excellent (30-30 cum laude) - Grade descriptors : Excellent knowledge of teaching contents for Integrated Course and laboratory activities; students should show high analytical and synthetic capabilities and should be able to apply their knowledge to solve problems. ECTS grade : B Very good (27-29) - Grade descriptors: Good knowledge of the teaching contents and excellent language control; students should show analytical and synthetic skills and be able to apply their knowledge of teaching contents and good language control; the student should be able to apply their knowledge to solve problems of medium and, in some cases, even higher complexity. ECTS grade: C Good (24-26)- Grade descriptors: Good knowledge of the teaching contents, in some cases limited to the main topic; acceptable ability to use the specific discipline language and independently apply the acquired knowledge. ECTS grade: E Su
TEACHING METHODS	Lectures, exercises and laboratory activities at the Regional Reference Center for Level II Laboratory Surveys and Quality Control (U.O.C. CQRC).

MODULE APPLIED PHYSICS

Prof. GAETANO SAVOCA

SUGGESTED BIBLIOGRAPHY

D. C. Giancoli, Fisica: principi e applicazioni, Casa Editrice Ambrosiana E. Ragozzino, Elementi di Fisica Per studenti di scienze biomediche, EdiSES			
AMBIT	10337-Scienze propedeutiche		
INDIVIDUAL STUDY (Hrs)	45		
COURSE ACTIVITY (Hrs)	30		
EDUCATIONAL OBJECTIVES OF THE MODULE			

Aquire the basic and general, principles of the mechanics, thermodynamics, optics and fluids physics, Acquire the ability to apply these physic laws to solve execises and examples

Hrs	Frontal teaching
2	Kinematics and Dynamics. Physical quantities. fundamental and derived quantities sizes. Units' measurement systems. Carriers. Vector and scalar. Sum of vectors. Product of vectors.
4	Kinematics of motion. material point. Speed. Acceleration. Trajectories and timing laws of motion. Motion in one and in two dimensions. Uniform rectilinear motion, uniformly accelerated motion, uniform circular motion, circular motion uniformly accelerated.
4	Dynamic. Force. Mass. gravitational forces. Force-Weight. Voltage of a rope. elastic forces. force of friction. Static and dynamic Attrition. Dynamics of circular motion.
6	Work and Energy. Work of a force. mechanical energy of a physical system. Kinetic energy. Potential energy. Forces and conservative systems. Difference between isolated and non- isolated systems. Conservation of mechanical energy. Potency and performance. Quantity 'of motion. Impulse of a force. Elastic and inelastic collisions
6	Fluids. Properties of liquid. Density. Specific weight. Stevin law. Pascal's Law. Archimede force. Fluid Dynamics. A fluid flow rate. Bernoulli's law. Biological and technical applications of the Bernoulli law.
4	thermodynamic bases. temperature. Thermometers based on espansion. clinical thermometer. Quantity of termic energy. Thermal Capacity of a body. Specific heat at constant volume or pressure of a substance. Heat, work and internal energy: first law of thermodynamics. metabolic power, energy value of foods. thermoregulation
4	Optics. luminous energy, refraction, reflection, diffraction, diffusion, frequency and wavelength, photoelectric effect.

SYLLABUS

MODULE BIOCHEMISTRY

Prof.ssa FRANCESCA DI GAUDIO

SUGGESTED BIBLIOGRAPHY

 Biochimica Medica, Siliprandi Tettamanti ed. Piccin

 Principi di Biochimica di Lenhinger, Nelson e Cox ed. Zanichelli

 AMBIT
 10338-Scienze biomediche

 INDIVIDUAL STUDY (Hrs)
 45

 COURSE ACTIVITY (Hrs)
 30

 EDUCATIONAL OBJECTIVES OF THE MODULE

Detailed knowledge of molecules of biologic interest and principal metabolic processes in which they are involved. Knowledge and understanding of enzyme action modality, regulation and catalysis. Knowledge and understanding of bioenergetic basis.

Hrs	Frontal teaching
3	Organic chemistry elements
2	Main food components
2	Basic cell functions
2	Membrane transport Simple and facilitated diffusion. Active transport. Glucose transporters (GLUT). Na+-glucose sinport. Na+/K+-dependent ATPase. Ca++ dependent ATPase. Other transport systems for calcium.
2	Amino acids and proteins
2	Role of the mineral component
2	Processes of catalysis, enzymes, enzymatic kinetics
2	Regulatory mechanisms, signal transduction
2	Protein metabolism
2	Glucid metabolism
2	Lipid metabolism
3	Nucleotide metabolism Purine e pyrimidine nucleotides biosynthesis. timidilic acid biosynthesis. Recovery nucleotides biosynthesis. Purine nucleotide catabolism Alterations of uric acid metaboolism. Hyperuricemia.
2	Vitamins General considerations. Features and functions of liposolubile and hydrosoluble vitamins. B1, B6 e B12 vitamins. Folic Acids. Monocarbon unit metabolism.
Hrs	Practice
2	Metabolic Diseases and Laboratory Evaluation of Metabolic Functionality

SYLLABUS