

UNIVERSITÀ DEGLI STUDI DI PALERMO

| DEPARTMENT | Scienze della Terra e del Mare | | | | |
|-------------------------|---|-------|------------------|--|-----------------------------|
| ACADEMIC YEAR | 2016/2017 | | | | |
| MASTER'S DEGREE (MSC) | ENVIRONMENTAL SCIENCES | | | | |
| INTEGRATED COURSE | GEOCHEMISTRY OF SURFACE PROCESSES - INTEGRATED COURSE | | | | |
| CODE | 18668 | | | | |
| MODULES | Yes | | | | |
| NUMBER OF MODULES | 3 | | | | |
| SCIENTIFIC SECTOR(S) | GEO/08, CHIM/02 | | | | |
| HEAD PROFESSOR(S) | CENSI PAOLO Professore Associato Univ. di PALERMO | | | Univ. di PALERMO | |
| OTHER PROFESSOR(S) | CENSI PAOLO Professore Associato Univ. di PALERMO | | Univ. di PALERMO | | |
| | MILIOTO | SIEFA | INA | Professore Ordinario | |
| | 12 | | | | |
| PROPAEDEUTICAL SUBJECTS | | | | | |
| MUTUALIZATION | | | | | |
| YEAR | 2 | | | | |
| TERM (SEMESTER) | Annual | | | | |
| ATTENDANCE | Not mand | atory | | | |
| EVALUATION | Out of 30 | | | | |
| TEACHER OFFICE HOURS | CENSI PA | OLO | | | |
| | Friday | 12:00 | 14:00 | Stanza nº 9 Via Archirafi, 36 2º | ° piano |
| | MILIOTO STEFANA | | | | |
| | Monday | 14:30 | 15:30 | Stanza 0/C9 - Dipartimento di Viale delle Scienze | Fisica e Chimica - Ed. 17 - |
| | Wednesday 14:30 15:30 | | 15:30 | Stanza 0/C9 - Dipartimento di Fisica e Chimica - Ed. 17 - Viale delle Scienze | |
| | Friday | 14:30 | 15:30 | Stanza 0/C9 - Dipartimento di Viale delle Scienze | Fisica e Chimica - Ed. 17 - |

| DOCENTE: Prof. PAOLO CENSI | |
|----------------------------|--|
| PREREQUISITES | chemistry, physics, mathematics, mineralogy and petrology |
| LEARNING OUTCOMES | Assessment capability of geochemical characters of water bodies and their relationships with human activities. Critical analysis of natural ecosystems with an interdisciplinary approach. Right exploitation of scientific language. |
| ASSESSMENT METHODS | oral examination of the approximate length of not more than 30 'articulated on questions involving the resolution of simple exercises that relate to concepts learned. During the course, students can take course tests aimed at establishing the lessons learned concepts and data manipulation techniques. |
| TEACHING METHODS | Lessons,numerical assessments, on-field trips and experiences |

MODULE SOLUTION PHASE PROCESSES

Prof. PAOLO CENSI

SUGGESTED BIBLIOGRAPHY

Ottonello G. - Principi di Geochimica - Zanichelli

materiali forniti dal docente (pubblicazioni scientifiche).

Huang, O'Melia and Morgan (1995) - Aquatic chemistry. American Chemical Society (Advance in Chemistry series 244).

Cotton S. - Lanthanide and Actinide chemistry - Wiley

| AMBIT | 50575-Discipline di Scienze della Terra |
|------------------------|---|
| INDIVIDUAL STUDY (Hrs) | 98 |
| COURSE ACTIVITY (Hrs) | 52 |

EDUCATIONAL OBJECTIVES OF THE MODULE

knowledge of fundamental parameters for the description of geochemical composition in a natural water body. Classification of natural waters according to the natural processes occurring therein and their influence on the Eh-pH variations.

| Hrs | Frontal teaching |
|-----|--|
| 5 | Classification of elements in relation with their behavior during primary and aqueous processes |
| 5 | dissolved elemental speciation |
| 2 | Eh-pH diagrams |
| 10 | Lanthanides and yttrium. the REE group. Y/Ho ratio. chemical and geochemical characteristics. |
| 8 | REE behavior during magmatic (CHARAC) processes. Dissolved REE behavior (non-CHARAC) processes |
| 10 | Geochemistry of Zr-Hf and Nb-Ta isovalent twins. Fractionation processes and the environmental significance of Zr/Hf and Nb/Ta ratios. |
| Hrs | Others |

MODULE

Field trip in interesting natural sites

FLUID-ROCK REACTIONS

Prof. PAOLO CENSI

SUGGESTED BIBLIOGRAPHY

12

Ottonello G. - Principi di Geochimica - Zanichelli

Huang, O'Melia and Morgan (1995) - Aquatic chemistry. American Chemical Society (Advance in Chemistry series 244). Cotton S. - Lanthanide and Actinide chemistry - Wiley

| AMBIT | 21017-Attività formative affini o integrative |
|------------------------|---|
| INDIVIDUAL STUDY (Hrs) | 51 |
| COURSE ACTIVITY (Hrs) | 24 |

EDUCATIONAL OBJECTIVES OF THE MODULE

Knowledge of the basic parameters that can describe the geochemical composition of natural water. Assess the extent of the changes in Eh, pH, natural water and the nature of the processes that determine it. Determine the mineralogical and textural effects induced by hydrolysis and weathering.

SYLLABUS

| Hrs | Frontal teaching |
|-----|---|
| 5 | Alteration processes of primary minerals and their transformation |
| 5 | Alteration of hydrothermal sulphides. Environmental effects of acid mine drainage induced by compositional changes of the surface water. |
| 5 | Water-rock interactions as a natural analogue of the leaching of solid waste. dependence their entities' from the mineralogical composition and texture |
| 9 | trace element co-precipitation in crystalline solids. A suitable remediation technique for trace elements in the ecosystem. |

SYLLABUS

MODULE PHYSICAL CHEMISTRY

Prof.ssa STEFANA MILIOTO

SUGGESTED BIBLIOGRAPHY

Chemistry of the Solid-Water Interface, W. Stumm, Wiley Principles of Colloid and Surface Chemistry, P. C. Hiemenz, Marcel Dekker. AMBIT 21017-Attività formative affini o integrative **INDIVIDUAL STUDY (Hrs)** 51 **COURSE ACTIVITY (Hrs)** 24

EDUCATIONAL OBJECTIVES OF THE MODULE

The aim of module consists into providing physico-chemical insights on the solid/water interphase highlighting the aspects related to surface charge. This will be done with the aim of giving a basis for the understanding of the aspects that will be studied in the other two modules, i.e. "Processes in solubilized phase" and "Reaction fluids-rocks".

| SYLLABUS | |
|----------|--|
| Hrs | Frontal teaching |
| 1 | Importance of the interphase. Surface and Interfacial tension. |
| 4 | Hydrated oxides/water interphace. Acid-basic behavior. |
| 6 | Surface charge and point of zero charge. Gouy-Chapman theory. Stern layer. |
| 2 | Effect of the pH and ionic strength on double electric layer |
| 2 | Zeta potential and its experimental determinations |
| 6 | Surface charge of silicates, carbonates, phosphates, clays and superficial complexes |
| 3 | Thermodynamics of adsorption at the solid/liquid interface. The Traube rule. |